Chemical Speciation, Bioavailability and Environmental Pollution Risks of Heavy Metals in the Greenhouse Soil Amended With Sewage Sludge and Municipal Solid Waste Compost

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Abstract—A study was carried out in intensive greenhouse areas of Antalya to evaluate the effects of repeated applications of sewage sludge (SS) and municipal solid waste compost (MSWC) on the metal bioavailability, metal-bound forms and pollution effects in greenhouse soil. SS and MSWC were applied to greenhouse soil for three cultivation years and total and DTPA-extractable Zn, Cu, Ni, Pb and Cd contents and metal speciation of greenhouse soil were determined. Additional to routine soil analysis, several environmental pollution indexes were used to evaluate the size of possible environmental pollution and risks.

SS and MSWC amendments increased both total and DTPA-extractable concentrations of Zn, Cu, Ni, Pb and Cd in the greenhouse soil when compared with the manure application. However, the concentration of heavy metals in greenhouse soils amended with SS and MSWC were found below the referenced limits. Relative availabilities of Zn, Cu and Pb in SS treatments were found greater than MSWC treatments. Compared to manure application, SS and MSWC treated soil often show significantly increased heavy metal contents, especially Zn, Cu, Pb, Ni and Cd. Repeated applications of SS and MSWC amendments were significantly increased bioavailable metals in the greenhouse soil compared to the manure. In sewage sludge and MSWC treatments, ‘total’ concentrations of all metals were found below the pollutant limits, but the increase in available fractions was more marked than those of total concentrations.

SS and MSWC amendments increased soluble, exchangeable and carbonate-bound fractions for all metals, especially Cu, Pb and Cd studied. MSWC applications to greenhouse soil caused higher mobility values of Zn, Ni, Pb and Cd than SS applications. Single-factor and composite Pollution Index values of greenhouse soil amended with manure, SS and MSWC were found below the critical value. Ecological risk factors for heavy metals were found below the critical level that indicate all metals posed low risk to surrounding ecosystem in short or medium-term. However, MSWC amendments increased risk assessment code values.

Keywords—Sewage sludge; Metals; Speciation, Bioavailability; Pollution risks.

I. INTRODUCTION

LAND disposal of sewage sludge is a major environmental problem. Increased urbanisation and industrialisation, especially in developing countries, require municipal authorities to handle larger volumes of sewage sludge often with limited resources. The use of sewage sludge in agriculture is now a widespread disposal practice.

Sewage sludge and MSWC contain valuable plant nutrients and organic matter that can improve soil fertility. The phytounritive capacity of compost has often been demonstrated to be analogous to that of manure; the same level of productivity, both quantitatively and qualitatively, can be maintained by replacing manure with compost [1, 2]. However, sewage sludge and MSWC often contains potentially toxic elements, that can cause soil contamination, phytotoxicity and undesirable residues in plant and animal products [3]. As a matter of fact, pollution problems may arise if toxic metals are mobilized may then occur through consumption of such crops and intake of contaminated waters. In the long term, the use of sewage sludge and MSWC can also cause a significant accumulation of Zn, Cu, Pb, Ni and Cd in the soil and plants [4].

Heavy metal pollution of agricultural soils and crops through the application of sewage sludge and MSWC as organic fertilizers is of great concern. At the present time, legislation in different countries limiting the use of sewage sludge in agriculture, refers to the total amounts of heavy metals in these wastes and in soils and recommends that soil pH to be maintained at 6 or higher. Nevertheless, these criteria are insufficient since mobility, environmental diffusion and bioavailability largely depend on soil physico-chemical characteristics and, likewise, on trace metal chemical forms [5]. From an environmental point of view, the evaluation and forecast of food contamination is related to the bioavailable fraction of heavy metals in soil. Long-term studies are needed to improve our understanding of the effects of land application of sewage sludge on soil chemical properties.
Today many environmental pollution risk indexes developed for water and sediments can be used for soils, organic matter and other environmental materials. Although several establishment criteria developed for soils depend on total concentrations, these criteria were frequently unsatisfied for a comprehensive environmental risk prediction. Due to potential toxicity of heavy metals to human life and environment, determining the chemical form of heavy metals in greenhouse soils is an important approach of chemical characterization and can provide useful information on its mobility and bioavailability.

The use of sewage sludge for agricultural purposes in Turkey is restricted, so far. Information on the fertilizing value of municipal solid waste composts and their effects on the heavy metal loading potentials on greenhouse soil are scarce. The aim of this study was to assess the effects of repeated applications of sewage sludge and MSWC on total and bioavailable (DTTPA-extractable) contents of Zn, Cu, Ni, Pb and Cd and on metal speciation and pollution risks of greenhouse soil.

II. MATERIAL AND METHODS

The experiment was conducted on the new constructed greenhouse representative of the major greenhouse growing area of Turkey, Antalya. The site studied is intensively cultivated and is not industrialized area. The geological materials of greenhouse area are mainly of calcareous nature and adjacent to Mediterranean sea with average 59 m altitude. The land is influenced by a Mediterranean climate with a high average annual rainfall (1038,8 mm/year), the annual average temperature being around 18.4 °C, 63.2 % average humidity and average 148.6 sunny days per year. As for greenhouses, the annual temperature is higher inside than outside, and most of them are watered by sprinklers with ground water source at the same point. All greenhouses have passive ventilation to control temperature and humidity inside. A great number of greenhouse soils is artificially built up with a different layer of sand, organic matter and other soil source.

The analytical characteristics of the greenhouse soil are shown in Table 1 which also shows the pollutant limits of soil permitted by EU legislation [6].

The experiment entailed the use of the following organic materials for the fertilization of the greenhouse soil.
- Composted cattle manure (CM): Produced by dairy-cows in sheds with straw bedding used for control treatment.
- Sewage sludge (SS): Sewage sludge was collected from the Hurma Waste Water Treatment Plant in Antalya. Sludge sample were air-dried and sieved through a 2-mm mesh and mixed well, respectively.
- Municipal solid waste compost (MSWC): Obtained from MSW composting plant in Kemer, Antalya. Compost was produced by the composting of the organic fraction of municipal solid waste, selected mechanically at the plant.

The mean analytical characteristics of organic materials tested are given in Table 2, which also shows the of EU limits [6] for sewage sludge.

Organic materials were applied to greenhouse soil for three cultivation years as an oven-dry basis at the following rates:
- Control: 50 ton ha\(^{-1}\) of cattle manure,
- SS\(_{50}\): 50 ton ha\(^{-1}\) of sewage sludge,
- MSWC\(_{50}\): 50 ton ha\(^{-1}\) of municipal solid waste compost.

### Table I: The Analytical Characteristics of the Experimental Greenhouse Soil Before Treatment

<table>
<thead>
<tr>
<th>Parameters</th>
<th>Characteristics</th>
<th>Limit values in soil, (mg kg(^{-1}) dry wt) [6]</th>
</tr>
</thead>
<tbody>
<tr>
<td>Texture</td>
<td>Loam</td>
<td></td>
</tr>
<tr>
<td>pH(\text{H}_2\text{O}) (1:5 w/v)</td>
<td>7.47</td>
<td>6.6(^{2}), 5.0-30.0(^{2})</td>
</tr>
<tr>
<td>CaCO(_3), %</td>
<td>7.70</td>
<td></td>
</tr>
<tr>
<td>Organic matter, %</td>
<td>2.04</td>
<td></td>
</tr>
<tr>
<td>Clay, % &lt;0.002 mm</td>
<td>15.5</td>
<td></td>
</tr>
<tr>
<td>CEC, cmol kg(^{-1})</td>
<td>9.5</td>
<td></td>
</tr>
<tr>
<td>EC, dS m(^{-1}) 25°C</td>
<td>0.8</td>
<td></td>
</tr>
<tr>
<td>Total N, %</td>
<td>0.12</td>
<td></td>
</tr>
<tr>
<td>P (ex), mg kg(^{-1})</td>
<td>13.7</td>
<td></td>
</tr>
<tr>
<td>K (ex), mg kg(^{-1})</td>
<td>96</td>
<td></td>
</tr>
<tr>
<td>Ca (ex), mg kg(^{-1})</td>
<td>1716</td>
<td></td>
</tr>
<tr>
<td>Mg (ex), mg kg(^{-1})</td>
<td>211</td>
<td></td>
</tr>
<tr>
<td>Zn, mg kg(^{-1})</td>
<td>86(^{3})</td>
<td>30.0-150.0(^{3})</td>
</tr>
<tr>
<td>Cu, mg kg(^{-1})</td>
<td>25</td>
<td>50-140</td>
</tr>
<tr>
<td>Ni, mg kg(^{-1})</td>
<td>16.6</td>
<td>30-75</td>
</tr>
<tr>
<td>Pb, mg kg(^{-1})</td>
<td>45</td>
<td>50-300</td>
</tr>
<tr>
<td>Cd, mg kg(^{-1})</td>
<td>0.002</td>
<td>1-3</td>
</tr>
</tbody>
</table>

\(^{1}\): Total concentrations, \(^{2}\): DTPA-extractable concentrations, \(^{3}\): Total concentrations at soil \(\text{pH}<7\).

The control treatment (cattle manure) is representative of the normal organic matter fertilization practice adopted in the provinces of Antalya. The rates of SS and MSWC were studied in order to introduce quantities of dry organic matter analogues to those of the control. Organic materials were manually incorporated into the greenhouse soil and mixed throughout the upper 20 cm. After the three cultivation season, greenhouse soil samples were taken at a depth of 20 cm and these were air-dried, sieved (< 2 mm) and stored in polyethylene bags sealed awaiting analysis.

### Table II: Average Analytical Characteristics of Organic Materials Used

<table>
<thead>
<tr>
<th>Parameters</th>
<th>CM Manure</th>
<th>SS</th>
<th>MSWC</th>
<th>Limit values [6], (mg kg(^{-1}) dry wt)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Moisture, %</td>
<td>54</td>
<td>38.6</td>
<td>35.4</td>
<td></td>
</tr>
<tr>
<td>pH(\text{H}_2\text{O}) (1:5 w/v)</td>
<td>7.96</td>
<td>6.45</td>
<td>7.66</td>
<td></td>
</tr>
<tr>
<td>Ash, %</td>
<td>26.2</td>
<td>44.2</td>
<td>48.5</td>
<td></td>
</tr>
<tr>
<td>Total N, %</td>
<td>2.12</td>
<td>1.88</td>
<td>1.15</td>
<td></td>
</tr>
<tr>
<td>P(_{2})O(_5), %</td>
<td>1.77</td>
<td>2.14</td>
<td>0.77</td>
<td></td>
</tr>
<tr>
<td>K(_{2})O, %</td>
<td>1.56</td>
<td>0.48</td>
<td>0.55</td>
<td></td>
</tr>
<tr>
<td>Total Zn, mg kg(^{-1})</td>
<td>180</td>
<td>1660</td>
<td>1220</td>
<td></td>
</tr>
<tr>
<td>Total Cu, mg kg(^{-1})</td>
<td>46</td>
<td>236</td>
<td>105</td>
<td></td>
</tr>
<tr>
<td>Total Ni, mg kg(^{-1})</td>
<td>16</td>
<td>54</td>
<td>43</td>
<td></td>
</tr>
<tr>
<td>Total Pb, mg kg(^{-1})</td>
<td>21</td>
<td>443</td>
<td>196</td>
<td></td>
</tr>
<tr>
<td>Total Cd, mg kg(^{-1})</td>
<td>0.06</td>
<td>2.8</td>
<td>1.6</td>
<td></td>
</tr>
</tbody>
</table>

http://dx.doi.org/10.15242/IJAAEE.ER1215048
Electrical conductivity (EC) and pH were measured as a soil:water ratio of 1:2. Cation exchange capacity (CEC) was determined by 0.1 M NaNO₃ extraction; CaCO₃ content was determined by the calcimeter; organic carbon was measured by wet oxidation; and texture was determined by Bouyoucos hydrometer method. For the determination of total heavy metal concentrations, soil, SS and MSWC samples were digested in aqua regia (1:3 HNO₃:HCl) according to the international standard [7]. Bioavailable fractions of metals were extracted from soil with diethylenetriaminepentaacetic acid-CaCl₂-triethanolamine adjusted to pH 7.3 (DTPA) procedure [8].

Sequential extraction method [9] was applied to soil samples to identify metal fractions. The heavy metal sequential extraction procedure had the following steps:
- F1. 1 M MgCl₂ (1:8 w/v, pH 7) for 1 h at room temperature; metals in soil solution and in exchangeable forms.
- F2. 1 M NaOAc (1:8 w/v, pH 5) for 5 h at room temperature; metals mainly in the carbonate fraction.
- F3. 0.04 M NH₄OH/HCl in 25 % (v/v) HOAc (1: 20 w/v) for 6 h at 96 °C; metals associated with Fe and Mn oxides.
- F4. 3 ml 0,02 M HNO₃+5 ml 30 % H₂O₂ (pH 2) for 3 h at 85 °C; metals associated with organic matter.
- F5. HNO₃–HCl digestion; residual fraction.

Total, bioavailable and sequential extracted metal concentrations were analysed using ICP-MS under optimised measurement conditions, and values were adjusted for oven dried (12 h at 105 °C) material.

Selected environmental pollution indexes for soil samples ‘Mobility of Metals’ (MF) [10], ‘Risk assessment code’ (RAC) [11], ‘Single-Factor and Composite Pollution Index of Soils’ [12] and ‘Potential Ecological Risk Factor Indexes’ [13] were used for comprehensive and integrated evaluation of possible pollution risks of treatments.

Statistical analyses were performed by using SPSS-16 for Windows program.

III. RESULT AND DISCUSSION

A. Total metal contents in the soil

Certain soil characteristics and heavy metal contents of experimental soil and their pollutant limits were given in Table 1. Experimental greenhouse soil has generally slightly alkaline reaction, moderate CEC, low EC values and highly calcareous, and having a lower agricultural fertility without fertilization. Average total metal contents were below the limits of European Union (EU), 86/278/CEE [6] directive to agricultural soils with pH>7. The heavy metal contents of organic materials studied (Table 2) are also well within the accepted normal range of values. A comparison of metal contents of organic materials with that of untreated soil showed that the metals Zn, Cu, Ni, Pb and Cd were present in both SS and MSWC in greater concentrations than in the soil. The three organic materials tested differed notably in their heavy metal content. Sludge metal concentrations are considerably higher than those of MSWC and cattle manure for all metals examined. The metal concentrations in cattle manure were the lowest. The heavy metal concentrations of all organic materials are below the levels indicated by the EU (86/278/CEE) for the agricultural use of sewage sludge.

The heavy metal contents of the greenhouse soil treated with SS and MSWC are given in Table 3.

![Table III: Total and DTPA extractable concentrations of heavy metals in the greenhouse soil](http://dx.doi.org/10.15242/IJAAEE.ER1215048)
and Cd- increase less consistently [17]. Based on the cumulative research in Europe into the agronomic use of compost, heavy metals tend to accumulate in soil and plants in the following order: Zn>Cu>Pb> Cd>Ni>Cr [18].

B. DTPA extractable metals in the soil

In this study, bioavailability of metals was expressed in terms of concentrations extractable with DTPA. All amounts of SS and MSWC brought about significant increases in DTPA-extractable Zn, Cu and Pb concentrations in comparison with the control. These results are in concordance with the data [19]. DTPA-extractable Zn, Cu and Pb also registered significantly higher values in 100 ton/ha than 50 ton/ha application level of SS and MSWC (Table 3). Although the content of Ni in ‘total’ form was increased by the application of SS and MSWC, DTPA-extractable form of Ni was unaffected. In the all treatments, DTPA-extractable Cd in the soil was always below the sensitivity of analytical method (Table 3).

The maximum permissible concentrations of heavy metals in surface soils amended with sewage sludge are normally based on total concentration, although it is the bioavailable metal fraction that poses environmental concern [20]. Total concentrations of all metals were found below the pollutant limits, but the increase in DTPA-extractable fractions was more marked than those of total concentrations.

The application of SS and MSWC caused to a significant increase in the extractability of metals in the soil. The relative increase of heavy metals Zn, Cu and Pb for SS and MSWC, in comparison with the control was higher for the DTPA-extractable than for the ‘total’ form. To compare the relative availability of different soil metals, DTPA/Total ratio is used to give available metals as a percentage of total soil metal. Relative availabilities, calculated in this way are given in Table 3. Relative metal availability, expressed as DTPA/Total ratios, was greater for Zn, Cu and Pb than the other metals in the greenhouse soil. Metals in control (manure) treatment were the least relative availability. Relative availability of Zn was the highest in SS treatment. Copper was more available in MSW treated plots. Relative availability of Pb was almost similar in SS and MSWC treatments. It should be noted that available metal may be quite large when the total metal present in the soil is great.

In the untreated soil (Table 1), DTPA solution extracted about 7.5 % of total Zn, 10.8 % of total Cu, 2.9 % of total Ni, 8.9 % of total Pb and 0.7 % of total Cd. DTPA-extractable Ni and Cd did not change significantly among the treatments, while DTPA-extractable Zn, Cu and Pb gradually increased in direct proportion to the amounts of SS and MSWC added to soil. These results are in concordance with the data [21], and support the hypothesis that metals added with sewage sludge may be more mobile in soil than native metals [22].

Simple correlation coefficients (r) between ‘total’ metals and DTPA-extractable metal content are reported in Table 4. With the exception of Cd (due to the unavailability of the relevant data) statistically significant and positive correlations were observed between ‘total’ and DTPA-extractable concentrations in the soil for all metals. Correlation coefficients between DTPA-extractable metals and ‘total’ metals were increased by the repeated applications of SS and MSWC for all metals except Cu.

### TABLE IV: SIMPLE LINEAR COEFFICIENTS BETWEEN DTPA-EXTRACTABLE METALS AND “TOTAL” METALS IN THE SOIL.

<table>
<thead>
<tr>
<th>Total metals</th>
<th>Zn</th>
<th>Cu</th>
<th>Ni</th>
<th>Pb</th>
<th>Cd</th>
</tr>
</thead>
<tbody>
<tr>
<td>DTPA-extractable metals</td>
<td>0.966 **</td>
<td>0.814 **</td>
<td>0.528 **</td>
<td>0.757 **</td>
<td>NS</td>
</tr>
</tbody>
</table>

*: Significant with P≤ 0.05; **: Significant with P≤ 0.01; NS: not significant.

C. Metal Speciation

Figure 1, 2 and 3 report the fractionation of heavy metals in greenhouse control soil and soil amended with SS and MSWC, respectively. For all metals, agreement between fractional total metal values and aqua regia extractable metal values are generally acceptable (100±5 %) rate. Concentrations of Zn, Cu, Ni, Pb and Cd in control soil fractions were given in Figure 1. The distribution of metals in greenhouse control soil samples generally followed the order below for the metals studied:

- Zn: F1<F3<F2<F4<F5
- Cu: F1≤F3<F2<F4<F5
- Ni: F1<F3<F2<F4<F5
- Pb: F1≤F3<F2<F4<F5
- Cd: F1<F2<F3<F4<F5

The study of the distribution of metals in control soil showed that the greatest percentage of all metals was present in the residual fraction (F5). F1, F2 and F3 fractions of Cd in control soil were higher than other metals. Amendment incorporation into the greenhouse soil significantly increased the sum of metal concentrations in the mobile fractions (Figure 2, 3). Chemical forms of metals in SS and MSWC may vary, depending on the solid-phase components present and their ability to release the metal, pH, temperature, number and accessibility of adsorption sites, metal affinity for solid components and operational parameters of the extraction process [23].

The distribution trends of Zn, Cu and Ni metals in SS and MSWC amendments were generally followed the F1<F3<F2<F4<F5 order. In MSWC amendments speciations of Cd metal was followed F3≤F4<F1≤F2<F5 the soil fractions. This property possibly gives Cd metal a high mobility. Despite the low total Cd concentrations in sludge and MSWC studied, the high solubility of Cd in the exchangeable phase indicates that this element could cause environmental damage and thus, the rate of sludge and MSWC applications should be taken into account [24].

The most mobile metal fraction in control soil was detected in Cd and the most immobile metal fractions were detected in Zn. Zinc largely associated with residual phase. However, an important portion of Cd was in labile form. SS and MSWC amendments increased F1, F2 and F3 fractions for all metals specially Cu, Pb and Cd studied. These results indicate the higher percentages of metals in soluble,
exchangeable and bound to Fe-Mn oxide fractions. The residual phase represents metals largely embedded in the crystal lattice of the soil fraction and should not be available for remobilization except under very harsh conditions [10].

D. Mobility of metals

Due to some metal forms are strongly bound to soil components than those extracted in F1, F3 and F5, the mobility of metals in soil samples may be evaluated on the basis of absolute and relative content of fractions weakly bound to soil component. Relative index of metal mobility was calculated as a ‘mobility factor’ (MF) [25] on the basis of the following equation:

$$ MF = \frac{(F_1 + F_2 + F_3)}{(F_1 + F_2 + F_3 + F_4 + F_5)} \times 100 $$

This equation is largely describes the potential mobility of metals. The MF values in control soil were considerably higher for Cd. SS and MSWC applications increased MF values for all metals. MSWC applications to greenhouse soil caused higher MF values of for all metals with the exception of Cu than SS applications. Copper was detected as the most mobile metal in SS amendments. The high MF values have been interpreted as symptoms of relatively high lability and biological availability of heavy metals in soils [25]. Using the upper baseline criterion, 95% of greenhouse soils present a relatively higher content of extractable heavy metals given their Cd and Cu concentration [26]. The results of the present study suggest that the mobility of the metals in control, SS and MSWC amendments declines in the following order, respectively: Cd>Cu>Ni>Zn>Pb, Cu>Pb>Cd>Zn>Ni and Cd>Pb>Cu>Zn>Ni (Figure 4).

E. Contamination evaluation of heavy metals

Single-factor and composite pollution index of soil: Single-factor pollution index (Pi) has been used to evaluate the degree of multi-element contamination. This is considered a better method of evaluation because heavy metals contamination in the surface environment is associated with a cocktail of contaminants rather than one element. Single indices are indicators used to calculate only one metal contamination:

$$ P_i = \frac{C_i}{S_i} $$

where $P_i$ is the single factor pollution index or contamination factor of heavy metal; $C_i$ is the measured contamination value of heavy metal, $S_i$ is the environmentally background contamination value of heavy metal.

Composite Pollution Index (PN) was applied to assess the quality of soil environment widely [12] and was defined as:

$$ P_{INemelov} = \sqrt{\frac{1}{m} \sum_{i=1}^{m} P_i^2 + \frac{P_{i, max}^2}{2} } $$

where $P_i$ is the single factor pollution index or contamination factor of heavy metal, $P_{i, max}$ is the maximum values of the single pollution indices of all heavy metals, $m$ is the count of the heavy metals species.

Single-factor and composite pollution coefficient of heavy metals are presented in Figure 5. It is clear that all contamination coefficients of metals in greenhouse soil were not exceeded critical value 1. Composite Pollution Index values of greenhouse soil amended with manure, SS and MSWC were also found below the critical value. Thus, all of soil samples may be considered as less contaminated by amendments and may be acceptable clean. The average Pi of heavy metals in greenhouse soils were ranked in the following order Zn<Ni<Cu<Pb>Cd.
Potential ecological risk factor indexes: An ecological risk factor (Er) to quantitatively express the potential ecological risk of a given contaminant [13] is:

$$ RI = \sum_{i=1}^{n} E_r^i \times C_f^i $$

where $T_r$ is the toxic-response factor for a given substance, and $C_f$ is the contamination factor, $n$ is the count of the heavy metal species. Although the risk factor was originally used as a diagnostic tool for the purpose of controlling water pollution, it was successfully used for assessing the quality of sediments and soils in environment by heavy metals.

The potential ecological risk assessment results of heavy metals in greenhouse soils are summarized in Figure 6. The average monomial risk factors, Er of heavy metals in greenhouse soils were ranked in the following order: Cd<Zn<Pb<Cu<Ni. In order to quantify the overall potential ecological risk of observed metals in the greenhouse soils, general ecological risk factor (RI) value was calculated as the sum of all the risk factors. The average monomial risk for heavy metals were found below the 40 that indicate all metals posed low risk to surrounding ecosystem in short or medium-term.

Risk Assessment Code (RAC): RAC classifies the risk levels based on the chemical speciation of heavy metals and reflects ecological risks. RAC determines the availability of heavy metals in sediments by applying a scale to the percentage of heavy metals in exchangeable and carbonate fraction [11]. In this study RAC parameter is modified to greenhouse soils. According to RAC guideline, classification of RAC is: <1%: no risk, 1-10 %: low risk; 11-30%: medium risk; 31-50%: high risk; >50 very high risk (Figure 7).

IV. CONCLUSION

SS and MSWC led to a far greater introduction of the heavy metals examined and brought about a significant increase in their ‘total’ form in the soil when compared with the control. However, the concentration of heavy metals in greenhouse soils amended with SS and MSWC were generally below the limits referenced by the 86/278/EEC directive to agricultural soils with pH >7.

The results of the present study indicated that soil application of SS and MSWC increased total and DTPA-extractable levels of Zn, Cu and Pb in the soil compared to the manure. Relative metal availability, expressed as DTPA/total ratios, was greater for Zn, Cu and Pb in SS than in MSWC treatments. Compared to manure application, SS and MSWC treated soil often show significantly increased heavy metal contents, especially Zn, Cu, Pb, Ni and Cd. Repeated applications of SS and MSWC in two years were significantly increased bioavailable metals in the greenhouse soil compared to the control.

SS and MSWC amendments increased soluble, exchangeable and carbonate-bound fractions for all metals, especially Cu, Pb and Cd studied. The most mobile metal fraction in control soil was detected in Cd and the most immobile metal fractions were detected in Zn. MSWC applications to greenhouse soil caused higher mobility values of for Zn, Ni, Pb and Cd than SS applications.

Single-factor and composite Pollution Index values of greenhouse soil amended with manure, SS and MSWC were found below the critical value. Ecological risk factors for heavy metals were found below the critical level that indicate all metals posed low risk to surrounding ecosystem in short or medium-term. However, MSWC amendments increased RAC values and Cd metal was detected in high risk level.
In this three-years study, no detrimental effects on soil properties were detected and metals in treatments were not higher than the allowed guideline level. However, taking into consideration the high potential bioavailability of heavy metals, repeated applications of SS and MSWC would carry a risk of progressive build-up of available trace elements in the soil in the course of time. For environmentally safe disposal of sewage sludge, immobilisation strategies for heavy metals in organic materials have to be investigated in the long term studies. Results demonstrate the importance of measuring extractable as well as total concentrations and also metal mobility and fractionation in topsoil when assessing likely effects on metal uptakes, and settings soil quality criteria.

REFERENCES