

Mineral Analysis of Milk through Atomic Absorption Spectroscopy and Their Biological Role in Human Life

Mehar Ali. Kazi

Abstract— Milk is essential human diet because it contains many essential trace minerals such as Calcium, Magnesium, Copper, Zinc, Sodium, Potassium and Phosphorous. Utilization of, milk is increasing at large scale through the world. It is essential to maintain the good quality of milk during production and manufacturing because presence of toxic metals in the milk becomes cause of health disturbance in human life. Many researchers identified the availability of heavy metals in the milk and it is necessary to know the level of heavy metal in the milk because of high utilization by infants and children's. The purpose of this research study is to analyze the presence of major and minor minerals in different milk samples. Flame Atomic Absorption Spectroscopy was used in the analysis of minerals

Keywords— Milk, Minerals, Flame Atomic Absorption Spectroscopy.

I. INTRODUCTION

THE milk is best diet for human health because it contains a good source of essential minerals such as calcium and phosphorous [1-3]. Due to the nutritional importance milk is consumed at large scale in recent time. Dairy products now days are considered as good source of nutrients for human health throughout the world [4-5]. Milk is also considered as raw material formed by animals [6-7]. Milk production should be controlled in the light of its component not only for protein and fats but also in the terms of mineral contents whenever possible. Milk is the best source of calcium and it can also provide some concentration of magnesium, zinc and some level of iron and copper [8-10]. Presence of phosphorous in the milk is essential for controlling acid base balance, energy metabolism and also organizes the cell membrane structure. Phosphorous works collaboratly with milk calcium and sustain the hardness of bones and teeth. To investigate the quality of food it is necessary to determine the concentration of trace metal contents such as calcium, phosphorous, magnesium, copper, iron and zinc etc. for analysis of these metals various techniques are applied like flame atomic absorption spectrophotometry, capillary zone electrophoresis. Many researchers identified the availability of heavy metals in the milk and it is essential to know the

level of heavy metals in the milk because of high utilization by infants and children [11].

II. EXPERIMENTAL

A. Reagent and Solution

All reagents were used of analytical grade (E. Merk). De-ionized water was used for the preparation of all the solutions. All glassware were first cleaned with detergent and distilled water, and then rinsed several times with de-ionized water. All glassware was left for dried for some time. All working standard solutions were prepared by diluting stock solution.

B. Sampling

Five different samples of milk were collected from Hyderabad super market and then samples were prepared for analysis on atomic absorption in nutritional research lab institute of biochemistry, University of Sindh, Jamshoro. These samples belong to different companies such as Tea Max (Haleeb Foods), Tarang (engro Foods), Milk-Pak (Nestle), City Sample, Village Sample.

C. Sample preparation

Take 5ml. of each milk sample in the volumetric flask as shown in Figure 1 and add 50ml. of 24% TCA for the digestion of milk sample. Quantities Shown in Table 1 and the digested samples are shown in Figure 1 Shake the samples at 5 minutes interval for 30 minutes, and then centrifuge each sample for 5 minutes at 4000 rpm. Filter each sample with filter paper and preserved the filtrate in the stopper bottle then forward it for the Atomic Absorption analysis (shown in figure 2.3).

TABLE I
QUANTITY FOR THE DIGESTION OF MILK

Milk Sample	Quantity (mL)	Amount of TCA (mL)
Tea Max	5	50
Tarang	5	50
Milk-pak	5	50
City sample (Hyd)	5	50
Village Sample	5	50



Fig.1. Milk samples in the volumetric Flask

III. RESULTS AND DISCUSSION

After the digestion of the samples, finally for the analysis samples were applied on Atomic absorption.. Before of the samples run on Atomic absorption, different parameters of A.A were settled like; Temperature and Pressure. Temperature was settled at the range of 2500 °C and Pressure of air at the range of 6Psi and the Pressure of Acetylene is 2Psi. After the settlement of parameters, standards of different ppm for every sample were applied first. After the standards, different digested samples of different ranges applied for analysis of the different minerals. Every sample gave different concentration of minerals one by one. The results obtained after analysis are given below in the concentration of ppm and milligram per deciliter (mg/dl).

TABLE II
CONCENTRATION OF CALCIUM IN MILK SAMPLES

Milk samples	Result in Ppm	Result in mg/dl	Nutritional Normal Range
Tea Max	38.45	76.9	1300 mg/day
Milk –Pak	29	58	1300 mg/day
Tarang	24.6	49.2	1300 mg/day
City Sample(Hyd)	40.36	80.72	1300 mg/day
Village Sample(umerkot)	20.6	41.2	1300 mg/day

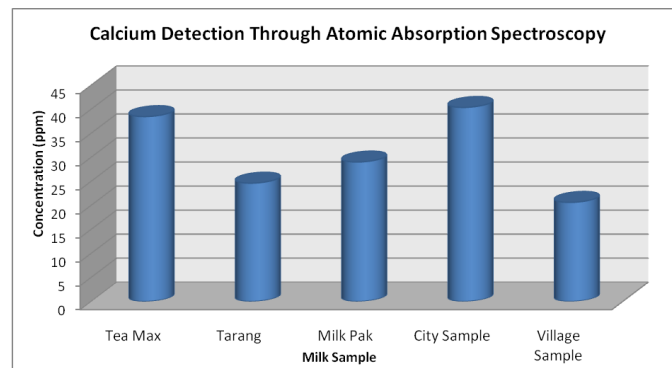


Fig.2. Graph of Calcium (Ca) Concentration shown in different Milk samples

TABLE III
CONCENTRATION OF COPPER IN MILK SAMPLES

Milk samples	Result in Ppm	Result in mg/dl	Nutritional Normal Range
Tea Max	0.012	0.024	0.9-1.3 mg/day
Milk –Pak	0.007	0.014	0.9-1.3 mg/day
Tarang	0.009	0.018	0.9-1.3 mg/day
City Sample(Hyd)	0.008	0.016	0.9-1.3 mg/day
Village Sample(umerkot)	0.007	0.014	0.9-1.3 mg/day

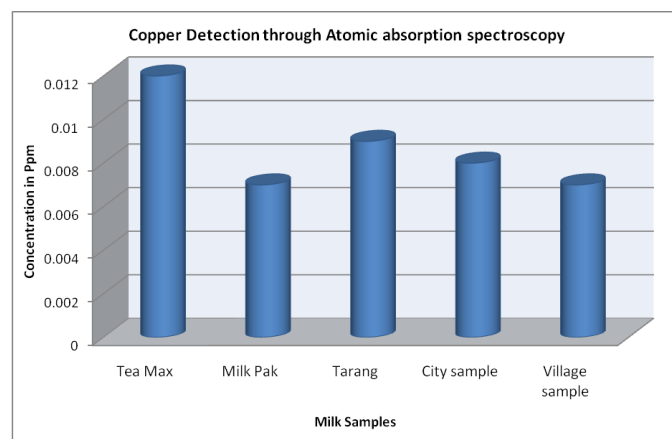


Fig. 3. Graph of Copper (Cu) Concentration shown in different Milk samples

TABLE IV
CONCENTRATION OF LEAD IN MILK SAMPLES

Milk samples	Result in Ppm	Result in mg/dl	Nutritional Normal Range
Tea Max	1.270	2.54	4-5 microgram/kg/day
Milk –Pak	0.725	1.45	4-5 microgram/kg/day
Tarang	0.816	1.632	4-5 microgram/kg/day
City Sample(Hyd)	0.698	1.396	4-5 microgram/kg/day
Village Sample(umerkot)	1.38	2.76	4-5 microgram/kg/day

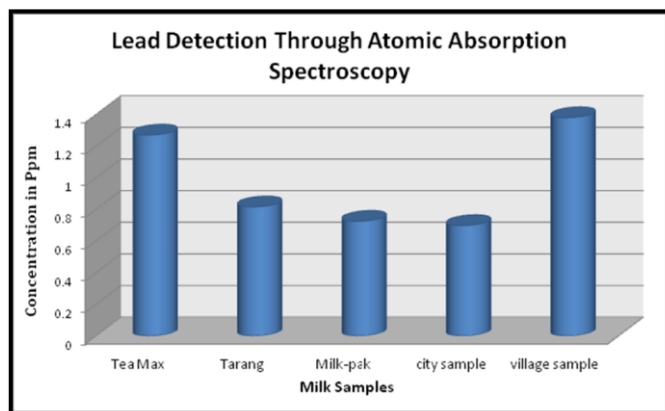


Fig. 4. Graph of Lead (Pb) Concentration shown in different Milk samples

TABLE V

CONCENTRATION OF LEAD IN MILK SAMPLES

Milk samples	Results
Tea Max	Not detectable
Milk -Pak	Not detectable
Tarang	Not detectable
City Sample(Hyd)	Not detectable
Village Sample(umerkot)	Not detectable

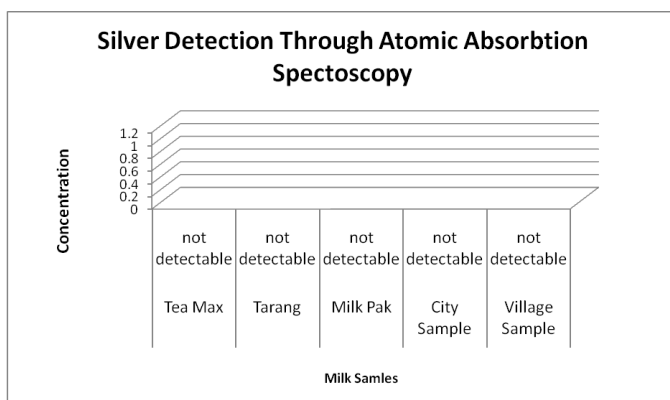


Fig.5. Graph of Silver (Ag) Concentration shown in different Milk samples

TABLE 3.5

COMPARISON AMONG DETECTED MINERALS FROM THE SAMPLES

Milk Samples	Calcium(Ca) in	Copper(cu) in ppm	Lead (Pb) in ppm	Silver (Ag) in Ppm
Tea Max	38.45	0.012	1.27	0
Milk-pak	24.6	0.009	0.816	0
Tarang	29	0.007	0.725	0
City Sample	40.36	0.008	0.698	0
Village sample	20.6	0.007	1.38	0

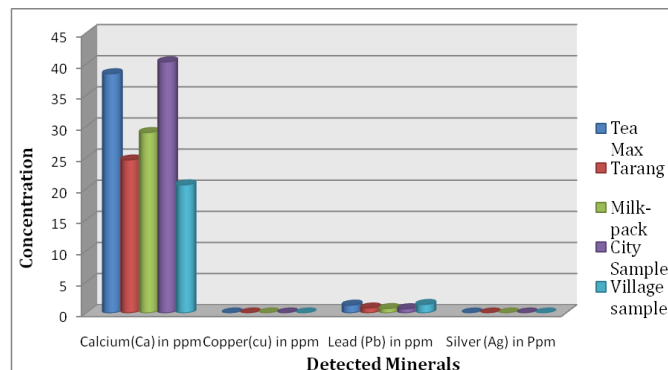


Fig.3.6. Comparison graph of detected Minerals from the samples

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